## From Graphs to Graphite!

Modelling Molecules with Graphs

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July 4, 2022

Ethane

## Why does $\mathrm{C}_{2} \mathrm{H}_{6}$ look like this?



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## Definition of a Graph

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The vertices are represented by points on the plane, and the edges by lines (not necessarily straight) joining the vertices.

## Some Terms Related to Graphs :)

- If an edge $e$ joins vertices $x$ and $y$ then $x$ and $y$ are adjacent and $e$ is incident with both $x$ and $y$.
- Any edge joining a vertex $x$ to itself is called a loop.
- Note that $E$ is a collection, not necessarily a set. This is to allow repeated edges. If two or more edges join the same two vertices, they are called multiple edges.
- A graph is said to be simple if it has no loops or multiple edges.
- The number of edges incident with a vertex $v$ in a graph without loops is called the degree or valency of $v$ and is denoted by $d(v)$.


## The Handshaking Lemma

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At a party, the total number of hands shaken is twice the number of handshakes. It has an immediate corollary: In any graph, the sum of the vertex degrees is even.

## Connected Graphs

Path: A sequence of edges which joins a sequence of vertices which are all distinct (and since the vertices are distinct, so are the edges).


## Connectedness

A graph is connected if, for each pair $x, y$ of vertices, there is a path from $x$ to $y$. A graph which is not connected is made up of a number of disconnected pieces, called components.

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"Pendant vertices": Vertex of degree 1.

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Then, let us call that leaf $I$.
Consider the path that starts at $I$, and follows to any other vertex $v$. If $v$ is a leaf, we are done. Otherwise it must be adjacent to at least another vertex $u$. Repeat the reasoning for $u$.

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Consider the path that starts at $I$, and follows to any other vertex $v$. If $v$ is a leaf, we are done. Otherwise it must be adjacent to at least another vertex $u$. Repeat the reasoning for $u$.
Thereby, our path can only end in a leaf or in an already visited vertex creating a cycle (which is not possible by assumption since the graph we consider is a tree).

## What is a "tree-like" molecule?

Let $T$ be a simple graph with $p$ vertices. Then the following statements are equivalent:

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2. $T$ has $p-1$ edges and no cycles.
3. $T$ has $p-1$ edges and is connected.

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1 \Longrightarrow 2 \Longrightarrow 3 \Longrightarrow 1
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- $T$ has an end vertex $w$.
- Remove $w$ and its incident edge from $T$ to obtain a tree $T^{\prime}$ with $k$ vertices.
- By the induction hypothesis, $T^{\prime}$ has $k-1$ edges; so $T$ has $(k-1)+1=k$ edges as required.


## Proof: If a simple graph $T$ with $p$ vertices has $p-1$ edges and no cycles then it is connected.

- Suppose $T$ has $p-1$ edges and no cycles, and suppose it consists of $t$ components, $T_{1}, \ldots, T_{t}$, each of which has no cycles and hence must be a tree.


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- Then $\sum_{i} p_{i}=p$, and the number of edges in $T$ is $\sum_{i}\left(p_{i}-1\right)=p-t$.
- So $p-t=p-1 \Longrightarrow t=1$, so that $T$ is connected.


## Proof: If a simple graph $T$ with $p$ vertices has $p-1$ edges and is connected then it must be a tree.

- Suppose $T$ is connected with $p-1$ edges, but is not a tree. Then $T$ must have a cycle.

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- The resulting graph must be a tree, with $p$ vertices and $q<p-1$ edges, contradicting the previous result.


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- The degree of a vertex is the "valency" of that atom- the number of atoms that it is connected with.
- In case an atom makes multiple bonds, its valency is reduced by one, per bond.


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| $a$ | $b$ | $q$ |
| :--- | :--- | :--- |
| 1 | 2 | 8 |
| 3 | 0 | 9 |

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1-propene

cyclopropane

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There is no edge connecting $v_{n-1}$ and $v_{n}$, which means there is no path from $v_{n-1}$ to $v_{n}$. This contradicts the fact that $G$ is connected.

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- As the molecule is a connected graph, number of edges $q \geq 3 n+2-1=3 n+1$.
- Thus $q=3 n+1=p-1$, which means that all our inequalities must attain the maximum. Thus all carbon atoms have degree four and the molecule is a tree.


## References

- A First Course in Discrete Mathematics, Ian Anderson.
- Introduction to Graph Theory, Douglas B. West.


## Thank You!

